50. The deposite on the filtre from the alkaline solution may then consist of mild stronthian, barytes, and calx, as also of argill, silex, and calx of iron.

51. To separate them, the whole, or all that is easily soluble, should be dissolved by digestion in nitrous acid; the digestion need not be obstinate, as only the argill, silex, and calx of iron, can make any considerable resistance, and they may afterwards be treated apart, as in No 34 and 41.

52. The barytic stronthian and calcareous earths being held in solution together, perhaps with some portions of argill and calk of iron, the two last should be got rid of by precipitation with caustic volatile alkali, and afterwards separated from each other by the means already often mentioned.

53. The folution is now supposed to contain only the three sirst earths; to ascertain this point it should first be well boiled, to expel the fixed air that might have been absorbed, and then a small portion of it essayed with stronthian lime water; if this produces a precipitate we may be assured that calx exists in the solution, otherwise not.

54. If stronthian lime-water produces no precipitation, we should essay another portion of the solution with barytic lime-water; if this produces a precipitate (stronthian lime-water producing none), we may infer that stronthian exists in the solution, and not calx; but, if barytic lime water produces none, we may conclude that barytes only exists in the solution; if both stronthian lime-water and barytic lime-water occasion precipitates, it remains dubious whether the three Hh44 earths