

Namen.	Formeln.	Gewicht eines Atoms.	+ E.	- E.	Wasser.
ACIDUM molybdic.	$\ddot{M}o^2$	1793,60			
	$\ddot{M}o^3$	2690,40			
- molybdosum	$Mo O^2 = \ddot{M}o$	796,80	74,90	25,10	
	$\ddot{M}o^2$	1593,60			
	$\ddot{M}o^3$	2590,40			
- mucicum .	$H^{10}C^6O^9 = \bar{M}u$	1318,32	C = 54,28	O = 60,68	H = 5,04
	$\bar{M}u^2$	2636,64			
	$\bar{M}u^3$	3954,96			
- muriatico - carbonicum	$\ddot{C}\ddot{M}$	617,98	$\ddot{C} =$ 44,55	$\ddot{M} =$ 55,45	
- muriatico - phosphoricum	$\ddot{P}\ddot{M}^6$	2605,55	$\ddot{P} =$ 54,25	$\ddot{M}^6 =$ 65,75	
- muriatico - phosphorosum	$\ddot{P}\ddot{M}^3$	1720,25	$\ddot{P} =$ 40,24	$\ddot{M}^3 =$ 59,76	
- muriatico - selenicum	$\ddot{S}e\ddot{M}^2$	1581,21	$\ddot{S}e =$ 50,58	$\ddot{M}^2 =$ 49,62	
- muriat. - sul- phuroso - car- bonicum	$\ddot{C}\ddot{M} + \ddot{S}\ddot{M}$	1361,79	$\ddot{M}\ddot{C} =$ 45,38	$\ddot{M}\ddot{S} =$ 54,62	
- muriaticum	\ddot{M}	342,65	41,63	58,37	
	\ddot{M}^2	685,30			
	\ddot{M}^3	1027,95			
- nitricum .	$NO^6 = \bar{N}$	677,26	11,41	88,59	
	$\bar{N}O^5$		N = 26,17	O = 73,83	
	\bar{N}^2	1354,52			
	\bar{N}^3	2031,78			
- nitrosum .	$NO^+ = \bar{N}$	477,26	16,19	83,81	