

Namen.	Formeln.	Gewicht eines Atoms.	+ E.	- E.	Wasser.
ACIDUM nitrosum	$\ddot{N}O^{\ominus}$		N = 37,14	O = 62,86	
	\ddot{N}^2	954,52			
	\ddot{N}^3	1431,78			
- oxalicum .	$HC^{+2}O^{+8} = 6O$	2710,17	C = 35,35	O = 66,41	H = 0,23
	\ddot{O}	451,69			
	\ddot{O}^2	903,40			
	\ddot{O}^3	1355,10			
- oximuriaticum	$MO^{\ominus} = \ddot{M}$	942,65	15,13	84,87	
	\ddot{M}^2	1885,30			
	\ddot{M}^3	2827,95			
- oximuriatos.	$MO^{\ominus} = \ddot{M}$	742,65	19,21	80,79	
	\ddot{M}^2	1485,30			
	\ddot{M}^3	2227,65			
- oxijodicum	$IO^{\ominus} = \ddot{I}$	2066,70	61,29	58,71	
	\ddot{I}^2	4153,40			
	\ddot{I}^3	6200,10			
- phosphoricum	$PO^{\ominus} = \ddot{P}$	892,50	45,97	56,05	
	\ddot{P}^2	1784,60			
	\ddot{P}^3	2676,90			
- phosphorosum	$PO^{\ominus} = \ddot{P}$	692,50	56,67	45,53	
	\ddot{P}^2	1384,60			
	\ddot{P}^3	2076,90			
- prussiacum	$C^2NH^2 = \ddot{P}$	339,56	C = 44,57	N = 51,72	H = 5,92
	\ddot{P}^2	679,12			
	\ddot{P}^3	1018,68			
- selenicum	$SeO^2 = \ddot{Se}$	695,91	71,26	28,74	
	\ddot{S}^2	1591,82			
	\ddot{S}^3	2087,75			